

Laser Plasma Aluminum Monoxide Emission Spectra Analysis

CHRISTIAN G. PARIGGER

Physics and Astronomy Department, University of Tennessee, University of Tennessee Space Institute, Center for Laser Applications, 411 B.H. Goethert Parkway, Tullahoma, TN 37388-9700, USA; cparigge@tennessee.edu

ABSTRACT: This investigation elaborates on analysis of aluminum monoxide, AlO, laser-plasma emission records using line strength data and the ExoMol astrophysical database. A nonlinear fitting program computes comparisons of measured and simulated diatomic molecular spectra. Predicted cyanide spectra of the AlO, $B^2 \Sigma^+ \longrightarrow X^2 \Sigma^+$, $\Delta v = 0, \pm 1, \pm 2, +3$ sequences and progressions compare nicely with 1 nanometer resolution experimental results. The analysis discusses experiment data captured during laser ablation of Al_2O_3 with 266-nm, 6-mJ pulses. The accuracy of the AlO line strength data is better than one picometer. This work presents as well comparison of the AlO line strength and of ExoMol data for the ²⁷Al¹⁶O diatomic molecule. Accurate AlO databases show a volley of applications in laboratory and astrophysical plasma diagnosis.

Keywords: diatomic molecules; aluminum monoxide; laser-plasma; data analysis; laser induced breakdown spectroscopy; combustion; time-resolved spectroscopy; spectra tting program; astrophysics

1. INTRODUCTION

The diatomic molecule aluminum monoxide, AlO, occurs in plasma-emission following laser ablation of aluminum containing samples, including alumina (Al_2O_3) [1] or aluminum containing alloys [2]. Combustion of aluminized propellants also reveals nice AlO ame emission spectra [3]. Usually, accurate diatomic line strengths data are preferred in the analysis [4-6] of recorded data. However, recent interest in exo-planet spectroscopy motivates determination of molecular databases, viz. ExoMol [7]. The ExoMol database lists various AlO isotopologues, however this work focuses on ${}^{27}Al^{16}O$. The transition of interest is the AlO $B {}^{2}\Sigma^{+} - X {}^{2}\Sigma^{+}$ band system that is similar in principle to previously communicated cyanide, CN $B {}^{2}\Sigma^{+} - X {}^{2}\Sigma^{+}$ band system [8].

Spectroscopy [9-15] of laser-plasma reveals clean AlO band system for delays of the order of several dozens of microseconds from the initial ablation plasma generation using pulse widths of a few nanoseconds. For aluminum monoxide spectroscopy, one can employ the ExoMol database in conjunction with the PGOPHER program for simulating rotational, vibrational and electronic spectra [16]. There are of course other databases that can be accessed [17] for diatomic molecules, including HITEMP that for example shows hydroxyl, OH, data [18]. The ExoMol AlO data files for the states and the transitions are converted in this work to line strength les for the purpose of utilizing previously communicated and extensively tested line-strength data that are freely available along with MATLAB [19] scripts for a subset of transitions associated with the AlO B-X band systems [4-6].

2. EXPERIMENTAL AND ANALYSIS OVERVIEW

The data from laser ablation experiments with frequency quadrupled Q-switched Nd:YAG radiation [1] show a range of 430 nm to 540 nm, and the published comparisons with line strength data reveal a temperature of 3,432 Kelvin at a delay of 20 µs. The measurements use standard laser-induced-breakdown-spectroscopy (LIBS) equipment.

International Review of Atomic and Molecular Physics, 14 (1), January-June 2023

The analysis in that work utilizes AlO-lsf line strengths and the Nelder-Mead downhill simplex, non-linear fitting algorithm [20]. The analysis communicated in this work is designed such that the same nonlinear method can be used, but the ExoMol data base for AlO is recast in a set of transitions with line strength data that are determined from Einstein A-coefficients.

A. Diatomic Molecular Analysis

The computation of diatomic molecular spectra utilizes line strength data. The Boltzmann equilibrium spectral program (BESP) and the Nelder-Mead temperature (NMT) program allow one to respectively compute an emission spectrum and fit theoretical to experimental spectra. The construction of the communicated molecular AlO line strengths "AlO-lsf" [5] first, makes use of Wigner-Witmer eigenfunctions and a diatomic line position fitting program, second, computes Frank-Condon factors and r-centroids, and third, combines these factors with the rotational factors that usually decouple from the overall molecular line-strength due to the symmetry of diatomic molecules. In turn, the ExoMol states and transition les for AlO [21, 22] are utilized for the generation of line strength data that can be used with BESP and NMT.

The ExoMol data show Einstein A-coefficients that are converted to line strengths [23-25], S, for electric dipole transitions, using

$$A_{ul} = \frac{16\pi^3}{3q_u h \epsilon_0 \lambda^3} \left(e \ a_0\right)^2 S_{ul}, \qquad g_u = 2(2J_u + 1).$$
(1)

Here, A_{ul} denotes the Einstein A-coefficient for a transition from an upper, u, to a lower, l, state, and h and ε_0 are Planck's constant and vacuum permittivity, respectively. The elementary charge is e, the Bohr radius is a_0 , and the transition strength is S_{ul} . The line strength, S, that is used in the MATLAB scripts is expressed in traditional spectroscopy units (stC² cm²). The wavelength of the transition is λ , g_u is the upper state degeneracy and J_u the total angular momentum of the upper state. In the establishment of line strength data, Hund's case (a) basis functions are preferred in connection with application of the Wigner and Witmer [26, 27] diatomic eigenfunction.

B. Air Wavelength vs. Vacuum Wavenumber

For NMT analysis the recorded, digital intensity values versus calibrated wavelength are utilized. The variation of the refractive index, r_i , of air at 15 °C, 101,325 Pa, and 0% humidity, with wavenumber [28],

$$10^8(r_i - 1) = \frac{k_1}{(k_0 - \sigma^2)} + \frac{k_3}{(k_2 - \sigma^2)},$$
(2)

where σ is the wavenumber in units of μm^{-1} , allows one to compute air wavelengths from the vacuum wavenumbers. Table I lists constants in Eq. (2).

Parameter	Value (μm^{-2})		
k_0	238.0185		
k_1	5,792,105		
k_2	57.362		
k_3	167,917		

 TABLE 1. Constants for variation of refractive index, see Equation 2.

3. RESULTS

This section elaborates analysis of recorded AlO spectra of the B ${}^{2}\Sigma^{+} \longrightarrow X {}^{2}\Sigma^{+}, \Delta v = 0, \pm 1, \pm 2, 3$ sequences and progressions. The use of ExoMol data and computed sets of line strength data that appear to be in use for extragalactic studies [7] would alleviate computation of specic transitions that are investigated in laser-plasma laboratory experiments. The ExoMol database shows 4,945,580 transitions and 94,862 states including the three lowest electronic states, X ${}^{2}\Sigma^{+}$, A ${}^{2}\Pi$, B ${}^{2}\Sigma^{+}$, C ${}^{2}\Pi$, D ${}^{2}\Sigma^{+}$, and E ${}^{2}\Delta$, e.g., 54,585 A states and 10,781 B states lead to 774,575 B-X transitions.

The AlO-lsf B-X data contain 33,484 transitions. The dierences in number of transitions are in part due to the number of rotational states, the cutos for Einstein A-coecients and associated line strengths (see Eq. (1)), or the establishment of sets of computed molecular parameters that fit data from high-resolution, Fourier-transform spectroscopy. The line positions are determined from high-resolution data with a standard deviation comparable to the estimated experimental errors of the high resolution line positions. The obtained, simulated line position accuracies are typically better than 0.05 cm⁻¹.

In this work, a Gaussian profile models the spectrometer and intensied linear-array detector transfer function. However, a measured system transfer function or a Voigt function can replace the selected Gaussian profile provided that changes are implemented in the MATLAB source scripts for the recently communicated BESP and NMT scripts [5]. N.B., the PGOPHER program allows one to accomplish Voigt profile fits.

A. Analysis with NMT Program and ExoMol Line Strengths

The AlO B-X line positions and Einstein A-coecients (that are converted to line strengths) are collected in a data file that is compatible with the mentioned NMT-spectral fitting program. Figure 1 illustrates spectra determined from temperature fitting with constant Gaussian line-width, $\Delta\lambda$. The simulated spectrum utilizes only AlO B-X transitions in the experimental range of 430 nm to 540 nm. Analysis of the measured data with the AlO-lsf data [1] reveals a temperature of T = 3,329 K, and a fitted FWHM of 1nm (43 cm⁻¹).





FIG. 1. (a) Experiment. (b) NMT fitting with ExoMol B-X data, T = 3,380 K, $\Delta\lambda$. = 1:0 nm.

B. Exomol AlO and AlO-lsf Data Comparisons

The simulated spectra are composed of quite a few individual rotational-vibrational transitions of the AlO B-X $\Delta v = 0, \pm 1, \pm 2, \pm 3$ sequences and progressions. Tables II and III summarize the number of lines in the data files.

TABLE 2. Number of B-X transitions and those in the experimental range 430 nm to 540 nm (18,500 cm⁻¹ to 23,250 cm⁻¹).

Database	AlO B-X	AlO B-X	
		430 nm to 540 nm	
ExoMol	774,575	169,143	
AlO-lsf	33,484	29,258	

Database	AlO B-X		
	$430~\mathrm{nm}$ to $540~\mathrm{nm}$		
	$\mathbf{A}_{\mathrm{coeff}} > 10^3 s^{-1}$		
ExoMol	104,260		
AlO-lsf	29,258		

Table IV displays agreements of lines within the indicated wavenumber range and otherwise the same identification for upper and lower levels of the transitions.

TABLE IV. Subset AlO B-X lines of the ExoMol data that agree within $\Delta \tilde{\nu}$ of 29,258 AlO B-X transitions in the AlO-lsf data for
the experiment range (18,500 cm⁻¹ to 23,250 cm⁻¹).

Database	$\Delta ilde{ u} <$	$\Delta ilde{ u} <$	$\Delta ilde{ u} <$	$\Delta ilde{ u} <$	$\Delta ilde{ u} <$	$\Delta ilde{ u} <$
	$0.05\mathrm{cm}^{-1}$	$0.2{\rm cm}^{-1}$	$1.0\mathrm{cm}^{-1}$	$2.0\mathrm{cm}^{-1}$	$10.0\mathrm{cm}^{-1}$	$20.0\mathrm{cm}^{-1}$
ExoMol	747	$3,\!146$	10,843	14,425	21,036	22,609

International Review of Atomic and Molecular Physics, 14 (1), January-June 2023

The differences in accuracy of the line positions can cause systematic errors in analysis of plasma emission spectra. Visualization of these dierences is suggested by (a) generating a "numerical experiment" spectrum using the AlO B-X data as extracted from the Exomol database, and then (b) analyzing the synthetic spectrum with the AlO-lsf database. Figure 2 exhibits the Exomol-database generated and AlO-lsf line strength data analyzed results.



FIG. 2. (a) Numerical experiment data, T = 3,380 K, $\Delta\lambda$ = 0.1 nm. (b) NMT fitting with AlO-lsf B-X data, inferred temperature from fixed line-width fitting: T = 3,200 K.

The obvious undulations in the dierence spectrum illustrates the ExoMol inaccuracies indicated in Table IV. A temperature of T = 3,380 K and a full-width-at-half-maximum, fixed Gaussian line-width, $\Delta \overline{\lambda}$, of 0.1 nm is selected for the "numerical experiment." Analysis by only fitting temperature yields T = 3,200 K, i.e., a temperature that is about five per cent lower than specied for the spectrum in Fig. 2 (a).

The AlO-lsf line strength database has been extensively tested [6]. The ExoMol database appears acceptable within $\simeq 20$ wavenumbers that corresponds to a spectral resolution of $\simeq 0.3$ nm. High-resolution experimental data may very well be accted by the inaccuracies of the line positions listed in the ExoMol database. The AlO-lsf database accuracy is better than 0.05 cm⁻¹ that corresponds to $\simeq 1$ picometer for the AlO B-X bands.

International Review of Atomic and Molecular Physics, 14 (1), January-June 2023

33

4. DISCUSSION

The AlO $B^2\Sigma^+ - X^2\Sigma^+$, $\Delta v = 0, \pm 1, \pm 2, +3$ sequences and progressions reveal many vibrational and rotational transitions that are usually not individually resolved in the study of laser-induced plasma emissions in the spectral range of 430 nm to 540 nm. Analysis of the 1-nm spectral resolution experimental emission spectrum with ExoMol line strengths and the NMT program shows AlO excitation temperature of $\simeq 3,380$ K that is consistent with previous analysis with AlO-lsf line strengths.

The agreement of the ExoMol AlO B-X and AlO-lsf line databases is marginal when using accuracies of the order of 0.05 cm⁻¹, or of the order of 1 picometer. However, for measurements with spectral resolutions of 43 cm⁻¹, or of the order of 1 nanometer, almost exactly identical results are inferred from tting of measured ablation spectra. A signicant advantage of the AlO-lsf database is its accuracy in predicting line position compared to the ExoMol database. The AlO-lsf line strength table is generated by fitting high resolution Fourier-transform data rather than computation from first principles. Analysis of predicted ExoMol spectra for dierent AlO isotopologues may be of interest in future research.

ACKNOWLEDGMENTS

The author (CGP) acknowledges the support in part by the Center for Laser Applications at the University of Tennessee Space Institute.

References

- [1] Dors, I.G.; Parigger, C.G.; Lewis, J.W.L. Opt. Lett. 23, 1778, (1998).
- [2] Surmick, D.M.; Parigger, C.G. Appl. Spectrosc. 68, 992 , (2014).
- [3] Parigger, C.G; Woods, A.C.; Surmick, D.M.; Donaldson, A.B.; Height, J.L. Appl. Spectrosc. 68, 362, (2014).
- [4] Parigger, C.G.; Hornkohl, J.O. Spectrochim. Acta Part A: Mol. Biomol. Spectrosc. 81, 404, (2011).
- [5] Parigger, C.G. Foundations 3, 1, (2023).
- [6] Parigger, C.G.; Hornkohl, J.O. Quantum Mechanics of the Diatomic Molecule with Applications; IOP Publishing: Bristol, UK, 2020.
- [7] Tennyson, J.; Yurchenko, S.N.; Al-Refaie, A.F.; Clark, V.H.J.; Chubb, K.L.; Conway, E.K.; Dewan, A.; Gorman, M.N.; Hill, C.; Lynas-Gray, A.E.; Mellor, T.; McKemmish, L.K.; Owens, A.; Polyansky, O.L.; Semenov, M.; Somogyi, W.; Tinetti, G.; Upadhyay, A.; Waldmann, I.; Wang, Y.; Wright, S.; Yurchenko, O.P. J. Quant. Spectrosc. Radiat. Transf. 255, 107228, (2020).
- [8] Parigger, C.G. Int. Rev. At. Mol. Phys. 14, 19 (2023).
- [9] Kunze, H.-J. Introduction to Plasma Spectroscopy; Springer: Berlin/Heidelberg, Germany, 2009.
- [10] Fujimoto, T. Plasma Spectroscopy; Clarendon Press; Oxford, UK, 2004.
- [11] Ochkin, V.N. Spectroscopy of Low Temperature Plasma; Wiley-VCH: Weinheim, Germany, 2009.
- [12] Demtröder, W. Laser Spectroscopy 1: Basic Principles, 5th ed.; Springer: Heidelberg, Germany, 2014.
- [13] Demtröder, W. Laser Spectroscopy 2: Experimental Techniques, 5th ed.; Springer: Heidelberg, Germany, 2015.
- [14] Miziolek, A.W., Palleschi, V., Schechter, I. (Eds.) Laser Induced Breakdown Spectroscopy (LIBS): Fundamentals and Applications; Cambridge Univ. Press: New York, NY, USA, 2006.
- [15] Singh, J.P.; Thakur, S.N. (Eds.) Laser-Induced Breakdown Spectroscopy, 2nd ed.; Elsevier: Amsterdam, The Netherlands, 2020.
- [16] Western, C.M. J. Quant. Spectrosc. Radiat. Transf. 186, 221, (2017).
- [17] McKemmish, L.K. WIREs Comput. Mol. Sci. 11, e1520, (2021).
- [18] Rothman, L.S.; Gordon, I.E.; Barber, R.J.; Dothe, H.; Gamache, R.R.; Goldman, A.; Perevalov, V.I.; Tashkun, S.A.; Tennyson, J. J. Quant. Spectrosc. Radiat. Transf. 2010, 111, 2139.
- [19] MATLAB Release R2022a Update 5, The MathWorks, Inc.: Natick, Massachusetts, US, 2022.
- [20] Nelder, J.A.; Mead, R. Comp. J. 7, 308, (1965).
- [21] Patrascu, A.T.; Yurchenko, S.N.; Tennyson J. Mon. Notices Royal Astron. Soc. 449, 3613, (2015).
- [22] Bowesman, C.A.; Shuai, M.; Yurchenko, S.N.; Tennyson, J. Mon. Notices Royal Astron. Soc. 508, 3181, (2021).
- [23] Condon, E.U.; Shortley, G.H. The Theory of Atomic Spectra; Cambridge Univ Press: Cambridge, UK, 1964.

- [24] Hilborn, R.C. Am. J. Phys. 50, 982, (1982).
- [25] Thorne, A.P. Spectrophysics, 2nd ed.; Chapman and Hall: New York, NY, US, 1988.
- [26] Wigner, E.; Witmer, E.E. Z. Phys. 51, 859, (1928).
- [27] Wigner, E.; Witmer E.E. On the structure of the spectra of two-atomic molecules according to quantum mechanics. In Hettema H. (Ed) Quantum Chemistry: Classic Scientic Papers. World Scientic: Singapore, SG, 2000; 287.



This document was created with the Win2PDF "print to PDF" printer available at http://www.win2pdf.com

This version of Win2PDF 10 is for evaluation and non-commercial use only.

This page will not be added after purchasing Win2PDF.

http://www.win2pdf.com/purchase/