Name____Key

February 14, 2001 SSN______Seat No_____

Complete the following equations. Name all organic reactants and products. (35%)

1-phenylcyclohexene

1-bromo-1-phenylcyclohexane

Br

b.
$$+ HNO_3 \xrightarrow{H_2SO_4} \longrightarrow NO_2$$

Benzene

Nitrobenzene

c.
$$H_3C$$
 H_2 H_3C H_4 H_5 H_5 H_5 H_6 H_7 H_8 H_8

1,1-diporpoxy-4-methylpentane

d.
$$CH_3-C$$
 $=$ $C-CH_3$ $\xrightarrow{1)HBr}$ $CH_3-\overset{CI}{C}-CH_2-CH_3$ \xrightarrow{Br} 2-bromo-2-chlorobutane CH_3 $\xrightarrow{CH_3}$ $\xrightarrow{CH_3}$ $\xrightarrow{CH_3}$ $\xrightarrow{CH_3}$ $\xrightarrow{CH_3}$ $\xrightarrow{CH_3}$

2-methylcyclooctanol

2-methylcyclooctanone

- 2. Draw the structures of the compounds whose names are given below. (20%)
- a. 2-Chloro-4-bromophenol

b. Trans-2-methylcyclopentanol

c. 1,4-diphenyl-3-pentanol

d. Diisopropyl ether

$$CH_3$$
 CH_3 CH_3 CH_3 CH_3

3. Write two equations which would allow the preparation of 2-butanol

3. Write two equations which would
$$\begin{pmatrix} H_2 \\ CH_3 & CH_3 \\ I \\ OH \end{pmatrix}$$
 in one step. (20%)

4. Write the mechanism for the reaction shown below. (15%)

5. The ir spectrum of an unknown compound shows it to be an alcohol, the mass spectrum gives a molecular weight of 60 and the carbon NMR shows two different kinds of carbons. Propose a reasonable structure for the unknown. (10%)